Semester	Code No	Course Title	Hours/ Week	Total Hours	Credit	Marks
1	PHY1C01	Complementary Course I: Properties of matter and Thermodynamics	2	36	2	75
	-	Complementary Course V: PHYSICS Practical	2	36	-*	-
2	PHY2C02	Complementary Course II: Optics ,Laser, Electronics	2	36	2	75
	-	Complementary Course V: PHYSICS Practical	2	36	-*	-
3	РНҮЗС03	Complementary Course III: Mechanics, Relativity, Waves and Oscillations	3	54	2	75
	-	Complementary Course V: PHYSICS Practical	2	36	*	-
4	PHY4C04	Complementary Course IV: Electricity ,Magnetism and Nuclear Physics	3	54	2	75
	PHY4C05	Complementary Course V: PHYSICS Practical	2	36	4*	100
Total					12	400

#### PHYSICS COMPLEMENTARY COURSE STRUCTURE Total Credits: 12 (Internal: 20%; External: 80%)

\* Examination will be held at the end of 4<sup>th</sup> semester

#### **COMPLEMENTARY COURSE THEORY: EVALUATION SCHEME**

The evaluation scheme for each course contains two parts: *viz.*, internal evaluation and external evaluation. Maximum marks from each unit are prescribed in the syllabus.

#### **<u>1. INTERNAL EVALUATION</u>**

20% of the total marks in each course are for internal evaluation. The colleges shall send only the marks obtained for internal examination to the university.

#### Table 1: Components of Evaluation

Sl. No.	Components	Marks for 2/3 credits papers
1	Class room participation based on attendance	3
2	Test paper: I	6
3	Assignment	3
4	Seminar/ Viva	3
	15	

#### Table 2: Pattern of Test Papers

Duration	Pattern	Total number of questions	Number of questions to be answered	Marks for each question	Mark s
	Short answer	12	10-12	2	20
2 Hours	Paragraph/proble m	7	6-7	5	30
	Essay	2	1	10	10
Total Marks*					60

\*90% and above = 6, 80 to below 90% = 5.5, 70 to below 80% = 5, 60 to below 70% = 4.5, 50 to below 60% = 4, 40 to below 50% = 3.5, 35 to below 40% = 3, 25 to below 30% = 2.5,15 to below 20=2, less than 15=0

#### **2. EXTERNAL EVALUATION**

External evaluation carries 80% marks. University examinations will be conducted at the end of each semester.

#### **Table 1: Pattern of Question Papers**

Duration	Pattern	Total number of questions	Number of questions to be answered	Marks for each question	Mark s
2 Hours	Short answer	12	10-12	2	20
	Paragraph/proble m	7	6-7	5	30
	Essay	2	1	10	10
Total Marks					60

## **Practical Evaluation** (Complementary)

Internal		External		
Record	4	Record with 20 experimrnts. Max. <sup>1</sup> / <sub>2</sub> mark for one expt.	10	
Regularity	4	Formulae, Theory, Principle	22	
Attendance	4	Adjustments, setting	14	
Test I	4	Tabulation & Observation	20	
Test II	4	Calculation, graph, result, unit	10	
		Viva	4	
Total	20	Total	80	

## **B.Sc. PHYSICS** COMPLEMENTARY COURSES SYLLABUS (For B. Sc Programme in Mathematics, Chemistry etc)

## Semester 1 Complementary course-I PHY1C01: Properties of matter & Thermodynamics

	Course Outcome	CL	KC	Class Sessions allotted
CO1	Understand the basic principles of elasticity	U	С	9
CO2	Understand the concepts of surface tension	U	С	5
CO3	Understand the aspects of viscosity	U	С	4
CO4	Understand the basic principles of thermodynamics	U	С	18

#### 36 Hours (Credit - 2)

#### Unit 1 Elasticity.

# Elastic modulii. (Elementary ideas)- Work done per unit volume - Poisson's ratio and theoretical limits - relation between various elastic constants(Derivation not required)- Twisting couple on a cylinder(Derivation not required)- Torsion pendulum-Determination of rigidity modulus of a wire-Bending of beams-bending moment- I-form girders- Cantilever loaded at the free end – Loaded uniformly (Derivation required )

#### Unit 2 Surface Tension & Viscosity.

Surface tension (Elementary ideas)-Excess pressure inside a liquid drop and bubble (Effect of electrostatic pressure on a bubble-change in radius)-Work done in blowing the bubble ( problem based on the formation of bigger drop by a number of smaller drops )

Viscosity-Coefficient of viscosity-Derivation of poiseuille's equation, stokes equation-Determination of viscosity by Poiseuille's method and stokes method-Brownian motion – Viscosity of gases

#### **Unit 3 Thermodynamics**

Thermodynamic processes –Indicator diagram (P-V diagram, P-T diagram, T-V diagram, T-S diagram )- Work done in Quasi static process-Work done in Isothermal, Adiabatic, Isochoric, Isobaric processes-First law of thermodynamics-Application to heat capacities- Second law of thermodynamics- Carnot's engine - Derivation of efficiency using Carnot's cycle-Carnot's theorem and its proof- Carnot's refrigerator( coefficient of performance )-

### 18 Hrs

9 Hrs

9 Hrs

**Entropy**- Change of entropy in a carnot's cycle, reversible cycle, irreversible cycleprinciple of increase of entropy- Entropy and available energy- entropy and disorder - Clausius-Clapyron equation(Derivation not required)-Effect of pressure on melting point and boiling point.

#### Text for study

- 1. Properties of matter-D. S. Mathur
- 2. Properties of matter-JC Upadhaya
- 3. Heat and Thermo dynamics- Brijlal and Subrahmanyam

#### **Books for reference**

- 1.. Heat and Thermo dynamics- D S Mathur
- 2. Heat and Thermodynamics Zemansky
- 3. Physics- Resnick and Halliday
- 4. Thermodynamics- Brijlal and Subrahmanyam

#### Mark distribution for setting Question paper.

Unit/ chapter	Title	Marks	
1	Elasticity	20	
2	Surface Tension & Viscosity	20	
3	Thermodynamics	39	
	Total Marks *	79	

\*Total marks include that for choice of questions in sections A, B and C in the question paper.

#### LAB PROGRAMME FOR COMPLEMENTARY COURSES

#### (Lab examination will be conducted at the end of 4<sup>th</sup> semester)

The minimum number of experiments for appearing examination is **75% of total 24 experiments** in the syllabus. Basic theory of the experiment must be shown at the time of Examination. **Students must submit a certified fair record at the time of Examination.** Number of Questions per session for the practical Examination shall be 8, and a minimum of 6 questions in the Question paper shall be set for the Examination at the centre.

## Semester 1 to 4 | Complementary Course V

#### PHY4C05: PHYSICS PRACTICALS I

	Course Outcome	CL	KC	Class Sessions allotted
CO1	Apply and illustrate the concepts of properties of matter through experiments	Ap	Р	36
CO2	Apply and illustrate the concepts of electricity and magnetism through experiments	Ap	Р	36
CO3	Apply and illustrate the concepts of optics through experiments	Ap	Р	36
CO4	Apply and illustrate the principles of electronics through experiments	Ap	Р	36

#### 36 Hours in each semester × 4 (Credit - 5)

#### List of Experiments

- 1. Characteristics of Diode and Zener diode
- 2. Liquid lens- Refractive index of liquid and glass
- 3. Torsion pendulum- Rigidity modulus
- 4. Spectrometer- Refractive index of the material of prism
- 5. Deflection Magnetometer- Moment of a magnet (Tan-A & Tan B positions)
- 6. Potentiometer-Measurement of resistance
- 7. Young's modulus Uniform bending -using optic lever
- 8. Static torsion Rigidity modulus
- 9. Spectrometer- Grating- Normal incidence
- 10. Melde's string- Frequency of fork (Transverse and Longitudinal mode)- (Mass determination
- by equal oscillation method / digital balance)

- 11. Half wave rectifier and Full wave rectifier
- 12. Field along the axis of a circular coil
- 13. Deflection Magnetometer- Moment of a magnet (Tan-C)

14. Potentiometer- Conversion of Galvanometer in to voltmeter –calibration by standard voltmeter

15. Viscosity of liquid- Capillary flow- Variable pressure head method (Mass determination by equal oscillation method / digital balance)

- 16. Logic gates Verification of truth table
- 17. Carey Fosters bridge- Resistivity of the material of wire
- 18. Surface Tension-Capillary rise method Radius by microscope.
- 19. Young's modulus of a cantilever- Pin and microscope method
- 20. Potentiometer-Calibration of low range voltmeter
- 21. Moment of inertia of fly wheel
- 22. Tangent galvanometer Reduction factor
- 23. Searle's vibration magneto meter Comparison of moments
- 24. Newton's rings- Wavelength of sodium light

#### **Books of Study:**

- 1. Electronics lab manual- K A Navas (vol 1 &2)
- 2. B.Sc Practical Physics- C L Arora

#### **Reference book:**

3. Practical Physics- S L Gupta & V Kumar